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| Mr. Bean Official Physical Chemistry for

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Law of Thermodynamics Laboratory

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book

Solutions, Suspensions, and ColloidsHow

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where T_0 is the freezing point of the solvent. If BC in Fig. 1st-1 is taken to be a straight line, the slope dp_A/dT may be replaced by CB/BD , i.e., by $(p_A - p_s)/\Delta T$, so that $p_p T H_p R T A s m v A 0 2 \Delta = \Delta \Delta 0$
Eq. 1st.4 The slope of the vapor pressure curve of the solid solvent is also given by another form of the

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where T_0 is the freezing point of the solvent If BC in Fig 1st-1 is taken to be a straight line, the slope dp_A/dT may be replaced by CB/BD , ie, by $(p_A - p_s)/\Delta T$, so that $p_p T H_p R T A s m v A 0 2 \Delta = \Delta \Delta 0$
Eq 1st4 The slope

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Physical chemistry laboratory manual - I.

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V.O.Chidambaram College, Tuticorin;

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Year 2 smoothly for a few minutes until
the temperature reaches a steady reading.
Record the temperature and the
equilibrium pressure. Increase the pressure
in the system by 3-5 mmHg (around 6
kPa) by allowing a small amount of air to
enter through tap B slowly.

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423/423A)
Chem 343 Physical Chemistry Laboratory
Manual Spring 2014 □ ADH 1/10/14 By
Audrey Dell Hammerich, Luke Hanley,
and Preston Snee with contributions from
Yoshitaka Ishii, Sanja Tepavcevic, and
Ben Stokes 1

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Table of Contents - Chemistry

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(Crockford, H.D., and Nowell, J.W.)

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(Crockford, H.D ...

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