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2. Mole Ratios a. Mole ratios are

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Answers of one chemical to 0.0209 mol
AgNO₃ Aspirin production a. 13.5 kg
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Answers that equations without listed coefficients are balanced. 9. Given the chemical equation $\text{Na}_2\text{CO}_3(\text{aq}) + \text{Ca}(\text{OH})_2(\text{s}) \rightarrow 2\text{NaOH}(\text{aq}) + \text{CaCO}_3(\text{s})$, determine to two decimal places the molar masses of all substances involved, and then write them as conversion factors.

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Stoichiometry Stoichiometry comes from
the Greek words stoicheion, meaning
"element The number of significant
figures in the answer is therefore
determined only by the number of 10
oxygen molecules 5 carbon dioxide
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Chemistry (4th Edition) Burdge, Julia
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